# The subject of the article is 6 tips for preventing rust.

The main idea of the text is to prevent rust.All over the world, people face many new and complex such problems every day. The rusting process begins when iron reacts with oxygen in the presence of water, saltwater, acids, or other harsh chemicals. As the iron oxide flakes off the metal surface, it exposes fresh iron molecules, which continue the reaction process. A ferrous metal is one that contains iron and only iron can rust. So 1 tip for preventing rust is this is to keep metal surfaces and air dry and clean, because oxygen in water molecules combines with iron to form iron oxide. It can also be noted that Iron needs to be protected from scratches, as scratches or cracks in the metal open up more metal and retain water. The next factor for protective coatings.Commercially available aerosol cans or cloths against rust can also protect metal objects, including tools, street gear, cars, and large metal parts. Stainless steel alloys contain iron, but it is resistant to rust because it also contains a high percentage of chromium, which is even more reactive than iron. Use Galvanized Metal. The zinc coating acts as a barrier to prevent oxygen and water from entering the steel. Rust spreads quickly, it is important to scrape it off as soon as it appears.