**1. Answer the questions to the text:**

1. What is [thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802" \o "THERMODYNAMICS)?

Thermodynamics is the study of the relationships between heat, work, and energy.

1. Where can [thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802) be applied?

It has a clear application to chemistry, biology, and other sciences.

1. How can physical life be described?

Physical life itself can be described as a continual thermodynamic cycle of transformations between heat and energy.

1. Are transformations perfectly efficient?

These transformations are never perfectly efficient.

1. Can the work output of a system be greater than the net energy input?

The work output of a system can never be greater than the net energy input

1. Is it possible to create a perpetual motion machine?

No, it is impossible.

1. What creations were made due to [the laws of thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82804)?

The laws of thermodynamics did make possible such highly useful creations as the internal combustion engine and the refrigerator.

1. How can any physical system be described?

Any physical system will spontaneously approach an equilibrium that can be described by specifying its properties, such as pressure, temperature, or chemical composition.

1. What do [the laws of thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82804) predict?

The three laws of thermodynamics predict the equilibrium state of the system.

**2. Insert a preposition or a conjunction if necessary:**

1. These transformations are never perfectly efficient, as the [second law of thermodynamics](http://www.answers.com/topic/second-law-of-thermodynamics-2) shows.
2. [Thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802) is the study of the relationships between heat, work, and energy.
3. The work output of a system can never be greater than the net energy input.
4. The three laws of [thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802) describe these changes and predict the equilibrium state of the system.
5. [The laws of thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82804) made possible such highly useful creations as the internal combustion engine and the [refrigerator](http://www.answers.com/topic/refrigerator).
6. It has a clear application to chemistry, biology, and other sciences.
7. It can be described by specifying its properties, such as [pressure](http://www.answers.com/topic/pressure), temperature, or chemical composition.
8. Any physical system will spontaneously approach an [equilibrium](http://www.answers.com/topic/dynamic-equilibrium).

**3. Insert a necessary word or word combination:**

1. [Thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802) is the study of the relationships between heat, work, and energy.
2. Any physical system will spontaneously approach an [equilibrium](http://www.answers.com/topic/dynamic-equilibrium) .
3. If external constraints are allowed to change, these properties generally change.
4. Many industrialists of the early nineteenth century believed it might be possible to create a perpetual motion machine.
5. Physical life itself can be described as a continual [thermodynamic cycle](http://www.answers.com/topic/thermodynamic-cycle) of transformations between heat and energy.
6. Physical system can be described by specifying its properties, such as [pressure](http://www.answers.com/topic/pressure), temperature, or chemical composition.
7. [The laws of thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82804) made possible such creations as the internal combustion engine and the [refrigerator](http://www.answers.com/topic/refrigerator).
8. The three laws of [thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802) describe these changes and predict the equilibrium state of the system.
9. The transformations are never perfectly efficient.
10. The work output of a system can never be greater than the net energy input.
11. [Thermodynamics](https://lms.kgeu.ru/mod/resource/view.php?id=82802) has a clear application to chemistry, biology, and other sciences.